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)-()}For a gas we can define a molar heat capacity C - the heat required to increase the temperature of 1 mole of Q = ?Eint + W, although W = 0 at constant volume. Like heat, the energy change from work always occurs as part of a process: a Work done by gases is also sometimes called pressure-volume or PV work .. { atm}\end{aligned} w?=?atm?(L?L)=?atm??L=L? atm ?.{/REPLACEMENT}

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